

Tetrahedron Letters, Vol. 35, No. 5, pp. 793-796, 1994 Elsevier Science Ltd Printed in Great Britain 0040-4039/94 \$6.00+0.00

0040-4039(93)E0309-8

## Palladium-Catalyzed Heteroarylation of 1-(tert-Butyldimethylsilyl)-3-indolylzinc Chloride. Efficient Synthesis of 3-(2-Pyridyl)indoles

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**Abstract:** Palladium-catalyzed coupling of 1-(*tert*-butyldimethylsilyl)-3-indolylzinc chloride (2) with either  $\pi$ -deficient or  $\pi$ -excedent heteroaryl halides gives 3-arylindoles in good to excellent yields. The method is extensively applied to the preparation of 3-(2-pyridyl)indoles.

Palladium catalyzed cross-coupling reactions to give 3-vinyl-, 3-alkynyl-, or 3-allylindoles from 3-indoleboronic acids, 3-indolyl triflates, 3-(chloromercurio)indoles, 3-haloindoles, or indole itself are well documented.<sup>1</sup> However only a very few procedures for the palladium-catalyzed arylation at the indole 3position are known.<sup>2</sup> In a recent paper,<sup>2a</sup> the cross-coupling reaction between 1-(benzenesulfonyl)-3indolylzinc iodide, prepared by oxidative addition of active zinc to the corresponding 3-iodoindole, and aryl halides to give 3-arylindoles in moderate to good yields has been described. These results prompted us to report our findings<sup>3</sup> about the heteroarylation at the indole 3-position, especially for the preparation of 3-(2pyridyl)indoles.<sup>4</sup>

As we recently reported,<sup>5</sup> 1-(benzenesulfonyl)-2-indolylzinc chloride, prepared by metathesis of 1-(benzenesulfonyl)-2-lithioindole with ZnCl<sub>2</sub>, efficiently reacts with 2-halopyridines in the presence of Pd(0) to give 2-(2-pyridyl)indoles. However, initial attempts to extend this procedure to the synthesis of 3-(2pyridyl)indoles from 1-(benzenesulfonyl)-3-indolylzinc chloride were unsuccessful due to the easy rearrangement of the intermediate 1-(benzenesulfonyl)-3-lithioindole to the 2-lithio isomer.<sup>6,7</sup>

Interestingly, we had observed<sup>8</sup> that 1-(*tert*-butyldimethylsilyl)-3-lithioindole is a stable species which does not rearrange to the 2-lithio isomer even upon warming at room temperature. For this reason we next decided to use the *tert*-butyldimethylsilyl group as the indole protecting group. The required 1-(*tert*-butyldimethylsilyl)-3-indolylzinc chloride (2) was prepared from 3-bromo-1-(*tert*-butyldimethylsilyl)indole<sup>8</sup> (1), by halogen-metal exchange with *tert*-BuLi at -78 °C followed by transmetallation (-78 °C  $\rightarrow$  rt) of the resulting 3-lithioindole with anhydrous ZnCl<sub>2</sub>. The reaction of 2 (1.5 equiv) with 2-bromopyridines **3a** and **3c** in the presence of 2 mol % of a catalyst prepared from PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> and DIBAH (2 equiv) in refluxing THF under argon for 4 h led to the respective N-protected 3-(2-pyridyl)indoles **4a** and **4c** in about 95% yield. Further treatment with a refluxing ethanolic solution containing a catalytic amount of *p*-TsOH yielded the

desilylated derivatives 5a and 5c in nearly quantitative yields. In all runs the corresponding 3,3'-biindole was formed in approximately 15% yield (based on 1).

In order to evaluate the scope of the reaction both the effect of substituents of a different nature on the pyridine ring and the use of heteroaryl halides derived either from other  $\pi$ -deficient or from  $\pi$ -excedent heterocycles were investigated. Thus, 2-bromopyridines 3e-g, having electron-releasing alkyl or methoxy substituents, and 2-chloropyridines 3i-m, having strongly electron-withdrawing nitro and ester groups, were subjected to the above cross-coupling conditions and the resulting crude mixtures immediately desilylated as described above to avoid partial desilylation during purification of 4 by column chromatography. As can be observed in Table 1, the overall yields for the desilylated 3-(2-pyridyl)indoles 5 were always excellent. The reaction is even compatible with acidic hydroxy substituents: compound 5h was obtained in 63% yield from 2-bromo-3-hydroxypyridine 3h.

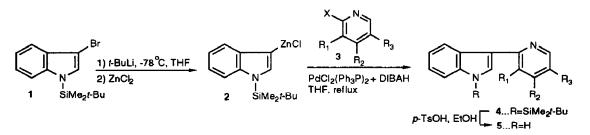


Table 1. The Palladium(0)-catalyzed Coupling of 3-Indolylzinc 2 with Halopyridines 3

	Halopyridine (3)				3-(2-Pyridyl)indole (5)	
	х	Ri	R <sub>2</sub>	R3	Product <sup>a</sup>	Yield(%) <sup>b</sup>
<u> </u>	Br	Н	Me	11	5a	93
3b	Cl	н	Me	11	5a	52
3c	Br	н	Н	11	5c <sup>C</sup>	95
3d	CI	н	н	11	5c	84
3e	Br	Et	Me	H	5e <sup>d</sup>	88
3f	Br	н	Me	Et	5f <sup>d</sup>	89
3g	Br	OMe	11	Н	5g	77
3h	Br	ОН	Н	H	5h	63
3i	Cl	CO2Me	н	П	5i	90
3ј	Cl	Ĥ	CO2Me	н	5j <sup>e</sup>	95
3k	Cl	н	Н	CO <sub>2</sub> Me	5kd	97
31	CL	н	н	NO <sub>2</sub>	51	89
3m	Cl	NO <sub>2</sub>	Н	NO2	5m	80

<sup>a</sup> All new compounds were identified by <sup>1</sup>II and <sup>13</sup>C NMR, IR, and elemental analysis. <sup>b</sup> Isolated yields (after column chromatography) based on 3. <sup>c</sup> Reference 4i. <sup>d</sup> Reference 4d. <sup>e</sup> Reference 4e.

On the other hand, treatment of 2-chloropyrazine with 3-indolylzinc chloride 2 under the same set of reaction conditions afforded 3-(2-pyrazinyl)indole 7a in 91% overall yield (Table 2). Similarly, halides derived from  $\pi$ -excedent heterocycles such as 2-bromothiophene, 3-bromothiophene, 3-bromofuran, and 3-bromo-(1-benzenesulfonyl)indole underwent cross-coupling reaction with 2 in the presence of Pd(0) to give the corresponding N-protected 3-(heteroaryl)indoles **6b-e**, although in lower yields. In these cases deprotection of indole had to be effected, after purification of 6, by treatment with tetrabutylammonium

fluoride (TBAF, 1.1 equiv, THF, 25 °C). Under these conditions, 3-(heteroaryl)indoles 7b-e were obtained in good yields.

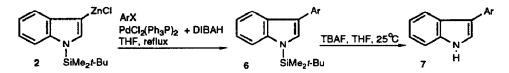


Table 2. The Palladium(0)-catalyzed Coupling of 3-Indolylzinc 2 with Heteroaryl Halides (ArX)

<sup>a</sup> All new compounds were identified by <sup>1</sup>H and <sup>13</sup>C NMR, IR, and elemental analysis. <sup>b</sup> Isolated yields (after column chromatography) based on ArX. <sup>c</sup> The bithiophene dimer was formed in ~25% yield. <sup>d</sup>Reference 10. <sup>e</sup> Reference 11.

The above results not only further illustrate the usefulness of the *tert*-butyldimethylsilyl protecting group in the chemistry of 3-lithioindoles<sup>8</sup> and constitute an extensive application of 3-indolylzinc derivatives,<sup>2a</sup> a novel class of indole metal derivative,<sup>9</sup> but also provide a simple, efficient, and general procedure for the synthesis of 3-(2-pyridyl)indoles, a structural unit present in a large number of indole alkaloids belonging to several structural types.

Acknowledgement: Financial support from the DGICYT, Spain (project PB91-0800) is gratefully acknowledged.

## **References and Notes**

 (a) Zheng, Q.; Yang, Y.; Martin, A. R. *Tetrahedron Lett.* 1993, 34, 2235. (b) Gribble, G. W.; Conway, S. C. Synthetic Commun. 1992, 22, 2129. (c) Sakamoto, T.; Kondo, Y.; Yasuhara, A.; Yamanaka, H. *Tetrahedron* 1991, 47, 1877. (d) Yokoyama, Y.; Ikeda, M.; Saito, M.; Yoda, T.; Suzuki, H.; Murakami, Y. *Heterocycles* 1990, 31, 1505. (e) Merlic, C. A.; Semmelhack, M. F. J. Organometal. Chem. 1990, 391, C23. (f) Hegedus, L. S.; Sestrick, M. R.; Michaelson, E. T.; Harrington, P. J. J. Org. Chem. 1989, 54, 4141. (g) Sakamoto, T.; Nagano, T.; Kondo, Y.; Yamanaka, H. Chem. Pharm. Bull. 1988, 36, 2248.
(h) Harrington, P. J.; Hegedus, L. S.; McDaniel, K. F. J. Am. Chem. Soc. 1987, 109, 4335. (i) Hegedus, L. S.; Toro, J. L.; Miles, W. H.; Harrington, P. J. J. Org. Chem. 1987, 52, 3319. (j) Yokoyama, Y.; Ito, S.; Takahashi, Y.; Murakami, Y. Tetrahedron Lett. 1985, 26, 6457. (k) Murakami, Y.; Yokoyama, Y.; Aoki, T. Heterocycles 1984, 22, 1493. (l) Itahara, T.; Kawasaki, K.; Ouseto, F. Synthesis 1984, 236. (m) Harrington, P. J.; Hegedus, L. S. J. Org. Chem. 1984, 49, 2657. (n) Itahara, T.; Ikeda, M.; Sakakibara, T. J. Chem. Soc., Perkin. Trans. I 1983, 1361. (o) Frank, W. C.; Kim, Y. C.; Heck, R. F. J. Org. Chem. 1978, 43, 2947.

- (a) Sakamoto, T.; Kondo, Y.; Takazawa, N.; Yamanaka, H. *Tetrahedron Lett.* 1993, 34, 5955. (b) Akita,
   Y.; Itagaki, Y.; Takizawa, S.; Ohta, A. *Chem. Pharm. Bull.* 1989, 37, 1477. (c) Ishikura, M.; Kamada,
   M.; Terashima, M. *Synthesis* 1984, 936.
- 3. This work was presented in a preliminary form at the Eighth European Symposium on Organic Chemistry (ESOC-8), Sitges, Spain, August 1993.
- Although several methods have been described for the synthesis of 3-(2-pyridyl)indoles, most of them require pyridine derivatives having electron-withdrawing substituents to achieve moderately good yields: (a) Seki, K.-I.; Ohkura, K.; Terashima, M.; Kanaoka, Y. Chem. Pharm. Bull. 1988, 36, 940. (b) Naito, T.; Iida, N.; Ninomiya, I. J. Chem. Soc., Perkin Trans. I 1986, 99. (c) Naito, T.; Iida, N.; Ninomiya, I. J. Chem. Commun. 1981, 44. (d) Suzuki, T.; Sato, E.; Goto, K.; Unno, K.; Kametani, T. Heterocycles 1980, 14, 433.(e) Kametani, T.; Suzuki, T. J. Chem. Soc. (C) 1971, 1053. (f) Kametani, T.; Suzuki, T. Chem. Pharm. Bull. 1971, 19, 1424. (g) Kametani, T.; Suzuki, T. J. Org. Chem. 1971, 36, 1291. (h) Hamana, M.; Kumadaki, I. Chem. Pharm. Bull. 1970, 18, 1742. (i) Powers, J. C. J. Org. Chem. 1965, 30, 2534.
- 5. Amat, M.; Hadida, S.; Bosch, J. Tetrahedron Lett. 1993, 34, 5005.
- 6. Saulnier, M. G.; Gribble, G. W. J. Org. Chem. 1982, 47, 757.
- 7. A similar result was reported in reference 2a.
- 8. Amat, M.; Hadida, S.; Sathyanarayana, S.; Bosch, J. J. Org. Chem., in press.
- (a) For a review on the use of transition metals in the synthesis and functionalization of indoles, see: Hegedus, L. S. Angew. Chem. Int. Ed. Engl. 1988, 27, 1113. (b) For a review on transition metal catalyzed reactions of organozinc reagents, see: Erdik, E. Tetrahedron 1992, 48, 9577.
- 10. Bergman, J. Acta Chem. Scand. 1971, 25, 1277.
- 11. Campbell, M. M.; Cosford, N.; Zongli, L.; Sainsbury, M. Tetrahedron 1987, 43, 1117.

(Received in UK 16 November 1993; accepted 26 November 1993)